

MT43

$\left\{ CF_2 - CF_2 \right\}_n$

Glass Filled Polytetrafluoroethylene (PTFE/Glass)

SPECIFICATIONS

| Property | Spec | Value |
|----------------------------------|------------|-------|
| Tensile Strength (psi) | ASTM D4745 | 2500 |
| Tensile Elongation (%) | ASTM D4745 | 230 |
| Hardness (initial; Shore D) | ASTM D2240 | 63 |
| Density (g/cm³) | ASTM D792 | 2.22 |
| Compressive Strength (psi) | ASTM D695 | 1325 |
| Compressive Modulus (psi x 10°) | ASTM D695 | 9.7 |
| Flexural Stress (lbs) | ASTM D790 | 10.5 |
| Flexural Strength (psi; maximum) | ASTM D790 | 2685 |
| Coefficient of Friction | | |
| Static | ASTM D1894 | 0.24 |
| Dynamic | ASTM D3702 | 0.33 |

DESCRIPTION

MT43 is a PTFE material with hardness 63 Shore D. specially compounded with glass filler. Polytetrafluoroethylene (PTFE) has exceedingly strong carbon-fluoride bonds (C-F). PTFE has a simple, linear, flexible and regular molecular structure, which makes it highly crystalline. Commercial PTFE is a high molecular weight polymer. Fluorine atoms form a tight sheath of protection providing PTFE with extreme molecular and physical properties. The sheath prevents PTFE from external influences upon the carbon-carbon backbone. It also results in weak interactions/bindings between polymer chains. These molecular structure properties make PTFE extremely resistant to chemicals or solvents even at very high temperatures and high pressures. PTFE also has very low friction and good anti-stick characteristics. PTFE is tough and flexible even at very low temperatures. However the same molecular structure properties result in mediocre mechanical properties with low stiffness and strength among thermoplastics. PTFE articles cannot be formed with conventional processes for thermoplastics because it does not flow above its crystalline melting point. Parts can be formed by a sintering process under high temperatures.